

REMARKS

In response to the Notice of Drawing Inconsistency with Specification (dated November 12, 2004) page 9 of the specification has been accordingly amended herein to include brief descriptions of drawing Figures 5, 6 and 7(a)-(c).

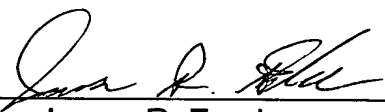
Basis for the brief description of Figure 5 is found at page 20, line 12 of the specification. Basis for the brief description of Figure 6 is found at page 8, lines 8-10 of the specification. Basis for the brief description of Figures 7(a), (b) and (c) is found at page 19, line 4 of the specification.

For convenience, and to expedite issuance of the present patent application, a replacement for page 9 is included in the Appendix herewith. The recitation at lines 15-16 of replacement page 9 as to "The examples below are illustrative examples in which all parts and percentages are by weight unless otherwise indicated" was previously inserted into page 9 of the specification in a Preliminary Amendment (dated June 20, 2001).

Also included in the Appendix herewith is a copy of the Notice of Drawing Inconsistency With Specification (dated November 12, 2004).

In light of the amendments herein and the preceding remarks, Applicants' specification is deemed to be in condition for issuance. Issuance of a patent at an early date is respectfully requested.

Respectfully submitted,

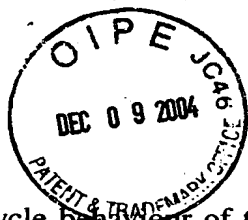
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APPENDIX

Replacement Page 9 of the Specification.

Copy of Notice of Drawing Inconsistency With Specification
(dated November 12, 2004)



In the drawings,

- Fig. 1 shows the cycle behaviour of the sample A obtained in the examples and of comparison samples E, F, N and V in the half-cell test;
- Fig. 2 shows the charge curve of sample A in the 10th cycle;
- 5 Fig. 3 shows the discharge curve of sample A in the 10th cycle; and
- Fig. 4 shows the x-ray diffraction spectrum of sample A.
- Fig. 5 shows an X-ray diffraction spectrum of Sample L of the Examples herein.
- Fig. 6 is a representative schematic of an electrolytic cell that may be used for the electrolytic preparation of nickel mixed hydroxide.
- 10 Fig.'s 7(a), (b) and (c) are SEM photographs of the powder of Example 12 herein.
- The nickel mixed hydroxide according to the invention is used preferably as a constituent of cathode materials in alkaline batteries, as in Ni/Cd or Ni/MH batteries, together with activators and auxiliaries known to the expert in the art.
- The invention is explained in more detail on the basis of the examples below.
- 15 The examples below are illustrative examples in which all parts and percentages are by weight unless otherwise indicated.